

Year Group	5	Term	Autumn 1	Subject	Science	Topic	Reversible and Irreversible Changes	
						Key Question	Are all changes reversible?	
Prior Learning and other Curriculum Links		and conthey observed and modegree identical descriptions of the control of the co	the properties an suggest so are used for. we that some when they are easure or reservature at whose Celsius (°C fy and compositing wood, metally of every o	me of the p materials of seheated or search the ich this hap ich the suite day material al, plastic, of and cardboar	change cooled, opensin ability of ls, glass, and for	Skills Statements	 Demonstrate that dissolving, mixing and changes of state are reversible changes Explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with burning and the action of acid on bicarbonate of soda Compare and group materials together, according to whether they are solids, liquids or gases Plan different types of scientific enquiries to answer questions, including recognising and controlling variables where necessary (Year 5 focus) Take measurements, using a range of scientific equipment, with increasing accuracy and precision, taking repeat readings when appropriate (Year 5 focus) Record data and results of increasing complexity using scientific diagrams and 	

			labels, classification keys, tables, scatter graphs, bar and line graphs (Year 5 focus) - Use test results to make predictions to set up further comparative and fair tests (Year 5 focus) - Report and present findings from enquiries, including conclusions, causal relationships and explanations of and degree of trust in results, in oral and written forms such as displays and other presentations (Year 5 focus) - Identify scientific evidence that has been used to support or refute ideas or arguments (Year 5 focus)
Fundamentals	 compare and grouptogether everyday materials on the basis of their properties, including their hardness, solubility, transparency, conductivity (electrical and thermal), and response to magnets know that some materials will dissolve in liquid to form a solution, anddescribe how to recover a substance from a solution use knowledge of solids, liquids and gases to decide how mixtures mightbe separated, including through filtering, sieving and evaporating give reasons, based on evidencefrom comparative and fair tests, for the 	Key Facts/Sticky Knowledge	 Reversible means it can be changed back to its original state Irreversible means that it cannot be returned to its original state and produces a new material. A change can occur through a change of temperature for example from hot to cold (melting) or cold to hot (heating) Irreversible changes produce a new material in a different state - e.g. rusting metal. Reversible changes produce a different state before returning to the old one, e.g. melting candle or evaporating water.

	particular usesof everyday materials,		A variable is the one thing you change in the				
	includingmetals, wood and plastic		test.				
	 demonstrate that dissolving, mixing and 		 A fair test is when everything is kept the 				
	changes of state are reversiblechanges		same except for the variable.				
	 explain that some changes result in the 						
	formation of new materials, and that						
	this kind of change is not usually						
	reversible, including changes associated						
	with burning and the action of acid on						
	bicarbonate of soda						
Our	They begin by learning about the differences between reversible and irreversible changes. They then						
Curriculum	perform a variety of experiments to identify reversible and irreversible changes - e.g. gas produced by						
Journey	bicarbonate soda and vinegar or rust on a nail. Finally, they use their understanding of irreversible changes to						
	perform their own experiement and create a change.						
Key	solid, liquid, soft, pour, flow, pile, pool,	Key	material, change, compare, contrast, solid, liquid,				
Vocabulary	surface, horizontal, runny, sticky, grain,	Vocabulary	gas, change of state, reaction, dissolve, melt,				
(revisited)	powder, ice, water, temperature, cool,	(new)	reversible,				
	cooling, warm, warming, hot, degree Celsius,		non-reversible				
	melt, melting, freeze, freezing, solidify,		Tion Total Sibio				
	solidifying, heating						